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Title: Determination of Metals by ICP-MS and ICP-OES (Optical Emission Spectrometry)		
Revision: 05	Replaces: CLG-TM3.04	Effective: 06/20/2016

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A. INTRODUCTION

1. Summary of Procedure

Sample is digested with concentrated nitric acid in a microwave digestion apparatus. The sample digest is diluted, fortified with internal standards, and analyzed using inductively coupled plasma mass spectrometry (ICP- MS) and inductively coupled plasma optical emission spectrometry (ICP-OES) depending on the metals being monitored.

2. Applicability

This method is suitable for quantification of the analytes listed below in the following species: beef, pork, poultry, and fish of the order Siluriformes (catfish; muscle only).

Table 1 - Applicable Analytes ICP-MS

Metal	Applicable Matrices	MLA(ppb)
Lead (Pb)	liver, kidney, muscle, processed products	≥ 25
Cadmium (Cd)	liver, kidney, muscle, processed products	≥ 10
Selenium (Se)	muscle, processed products	≥ 500
Manganese (Mn)	muscle, processed products	≥ 200
Molybdenum (Mo)	muscle, processed products	≥ 50
Thallium (Tl)	muscle, processed products	≥ 50
Cobalt (Co)	muscle, processed products	≥ 25

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Table 2 - Applicable Analytes ICP-OES

Metal	Applicable Matrices	MLA (ppm)
Iron (Fe)	muscle, processed products	≥ 30
Zinc (Zn)	muscle, processed products	≥ 30
Copper (Cu)	muscle, processed products	≥ 3
Nickel (Ni)	muscle, processed products	≥ 6
Aluminum (Al)	muscle, processed products	≥ 24
Boron (B)	muscle, processed products	≥ 4.8
Barium (Ba)	muscle, processed products	≥ 3.6
Chromium (Cr)	muscle, processed products	≥ 3.6
Vanadium (V)	muscle, processed products	≥ 3.6
Strontium (Sr)	muscle, processed products	≥ 3

Note: Refer to 21CFR for tolerance values set by FDA and 40CFR for tolerance values set by EPA.

B. EQUIPMENT

Note: Equivalent equipment may be substituted.

1. Apparatus

- a. Analytical balance - sensitive to 1 mg, Mettler, PG403-S
- b. MarsXpress Microwave and MarsXpress Digestion System, CEM.

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- c. MarsXpress Microwave Digestion vessels - 55 mL capacity liners, vent plugs, and screw caps, CEM, cat. nos. 574125 and 212020.
 - d. Vacuum Concentration/Drying apparatus - Microvap accessory set for MarsXpress system, CEM.
 - e. Vacuum Scrubber Module and Dual Head Vacuum Pump, 120V/60Hz, CEM.
 - f. Stirring rods (optional) - Teflon or polypropylene, Lab Depot, cat. no. F377390001.
 - g. Volumetric flasks - polypropylene or polymethylpentane, 50, 100, 1000 mL, class A, Lab Depot cat. no. 5460P-50, 5640P-100, 5640P-1L .
 - h. Volumetric flasks - glass, 10 - 1000 mL, as needed for preparation of standards, reagents class A, VWR cat. no. 89000-398 (10 mL) - 89000-412 (1000 mL).
 - i. Micropipettors - fixed or variable, covering ranges 10 - 5000 μ L.
 - j. Bottles - polypropylene, 100 and 250 mL, Fisher, cat no. 02-893A, and 02-893B.
 - k. Centrifuge tubes - polypropylene, 50 mL, Fisher Scientific, cat. no. 06-443-18.
 - l. Argon gas, high purity grade (99.99%).
 - m. Syringe filter (optional) - Acrodisc CR 13 mm, with 0.2 μ m PTFE Membrane, Gelman Laboratory, VWR, cat no. 28143-982.
 - n. Milestone Trace Clean (optional).
 - o. Helium gas, ultra high purity \geq 99.999%
 - p. Hydrogen gas, ultra high purity \geq 99.999%
2. Instrumentation
- a. Inductively Coupled Plasma Mass Spectrometer - Agilent model 7500ce, equipped with an Octapole Reaction System (ORS).
 - b. Inductively Coupled Plasma Spectrometer - Perkin-Elmer, model 5300DV.

C. REAGENTS AND SOLUTIONS

Note: Equivalent reagents / solutions may be substituted. The stability time frame of the solution is dependant on the expiration date of the components used. The maximum length of time that a working reagent shall be used is 1 year unless the laboratory has produced extension data.

1. Reagents
- a. Deionized water (DI water) - for cleaning only.
 - b. Millipure water - Deionized water polished to ASTM CAP/NCCLS Type 1 specifications or better (resistance \geq 18 megaohms).

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- c. Nitric acid (HNO₃) - concentrated. Ultra-pure grade (Optima by Fisher or Double Distilled by GFS) recommended. For samples, standards and reagents.
- d. Nitric acid (HNO₃) - concentrated. Metal analysis or ICP-OES grade (e.g., OmniTrace by EMD). For cleaning only.
- e. Sodium Hydroxide (NaOH) - reagent grade.
- f. Mass spectrometer tuning solution (10 µg/L) - Lithium, Yttrium, Cerium, Thallium, and Cobalt in 2% HNO₃, Cat No. 5184-3566, Agilent Technologies.

2. Solutions

- a. 25 - 50% NaOH solution (for evaporation scrubber):
Weigh 250 - 500 g of NaOH into a container with graduations and bring the volume up to approximately 1 L with water and mix.
- b. 2% HNO₃ solution:
Dilute concentrated HNO₃ 1:50 with millipure water (e.g., 20 mL/1L). Prepare and store in polypropylene bottles.
Note: Additional HNO₃ concentrations may be necessary based on commercial standards.

D. STANDARD(S)

Note: Equivalent standards / solutions may be substituted. Purity and counterions are to be taken into account when calculating standard concentrations. The stability time frame of the solution is dependant on the expiration date of the components used. In-house prepared standards shall be assigned an expiration date that is no later than the expiration date of the earlilest expiring component or no later than the stability stated in the method, whichever ends soonest. The maximum length of time that an in-house prepared standard shall be used is 1 year unless the laboratory has produced extension data.

Note: All standards and solutions may be stored at room temperature.

1. Standard Information

All elemental standard and internal standard solutions are prepared from commercial reference standards, which are available at concentrations of 1,000 or 10,000 mg/L (µg/mL). Reference standards must be ICP-MS grade.

- a. Elemental standard solutions

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Table 3 - Analytical Standard Information - ICP-MS

Company	Name	Catalog No.
Inorganic Ventures, Lakewood, NJ	Cadmium	CGCD10-1
Inorganic Ventures, Lakewood, NJ	Cobalt	CGCO10-1
Inorganic Ventures, Lakewood, NJ	Lead	CG-PB10-1
Inorganic Ventures, Lakewood, NJ	Manganese	CGMN10-1
Inorganic Ventures, Lakewood, NJ	Molybdenum	CGMO10-1
Inorganic Ventures, Lakewood, NJ	Selenium	CGSE10-1
Inorganic Ventures, Lakewood, NJ	Thallium	CGTL10-1
SCP Science, Champlain, NY	Cadmium	140-061-480
SCP Science, Champlain, NY	Cobalt	140-061-270
SCP Science, Champlain, NY	Lead	140-061-820
SCP Science, Champlain, NY	Manganese	140-061-250
SCP Science, Champlain, NY	Molybdenum	140-060-420
SCP Science, Champlain, NY	Selenium	140-061-340
SCP Science, Champlain, NY	Thallium	140-061-810

Table 4 - Analytical Standard Information - ICP-OES

Company	Name	Catalog No.
SCP Science, Champlain, NY	Iron	140-061-260
SCP Science, Champlain, NY	Zinc	140-061-300
SCP Science, Champlain, NY	Copper	140-061-290
SCP Science, Champlain, NY	Nickel	140-061-280
SCP Science, Champlain, NY	Aluminum	140-061-130
SCP Science, Champlain, NY	Boron	140-060-050
SCP Science, Champlain, NY	Barium	140-061-560
SCP Science, Champlain, NY	Chromium	140-061-240
SCP Science, Champlain, NY	Vanadium	140-061-230
SCP Science, Champlain, NY	Strontium	140-061-380

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Table 5- Internal Standard Information

Company	Name	Catalog No.
SCP Science, Champlain, NY	Bismuth	140-061-831
SCP Science, Champlain, NY	Gallium	140-061-311
SCP Science, Champlain, NY	Germanium	140-060-320
SCP Science, Champlain, NY	Indium	140-061-491
SCP Science, Champlain, NY	Rhodium	140-062-451
SCP Science, Champlain, NY	Scandium	140-061-210
SCP Science, Champlain, NY	Terbium	140-061-651
SCP Science, Champlain, NY	Yttrium	140-061-391

2. Preparation of Standard Solution(s)

Important: Metals may be leached from glass by nitric acid. Store all standard solutions in polypropylene or other inert containers. If glassware is used, it should be cleaned with nitric acid and dedicated for trace metals analyses. Standards prepared in glassware should be used immediately or transferred to suitable containers for storage.

a. Internal standard (ISTD), (5000 µg/L):

Add volumes of reference standard solution(s) (see table 11) equivalent to 500 µg (e.g., 500 µL of a 1000 mg/L solution) to a 100 mL volumetric flask and dilute to 100 mL with 2% HNO₃. Mix.

Note: Other internal standards can be used as long as the element is not contained in the sample, the mass number is similar to that of the analyte, and the ionization potential is similar to that of the analyte.

b. Calibration Standards - ICP-MS

Calibration standards are required for constructing a multipoint standard curve covering the range of analyte concentrations anticipated in samples.

Prepare intermediate standards by making dilutions of commercially available standard solutions into a concentration of HNO₃ similar to the commercial solution. Suggested concentrations, based on use of 1000 mg/L standards, are:

i. 10,000 µg/L:

Pipet 100 µL of 1000 mg/L standard to a 10 mL volumetric flask and dilute to volume.

ii. 1000 µg/L:

Pipet 10.0 mL of 10,000 µg/L solution (i) to a 100 mL volumetric flask and dilute to volume.

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iii. 100 µg/L:

Pipet 1.00 mL of 10,000 µg/L solution (i) to a 100 mL volumetric flask and dilute to volume.

Prepare calibration standards by making appropriate dilutions of intermediate standards with 2% HNO₃ and adding sufficient 5000 µg/L ISTD to result in a final ISTD concentration of 5 µg/L. Prepare these standards using polymeric volumetric flasks.

The Table below lists some suggested concentrations for calibration standards and recommended volumes and concentrations of solutions required for preparation of 100 mL volumes of each.

Table 6 - Calibration Standards - ICP-MS

Calibration STD Conc. [Sample Conc* in ()]	Amount used x Intermediate Standard concentration	Amount ISTD
Calibration Blank (0 ppb)	2% HNO ₃ Only	100 µL
0.05 µg/L (5 ppb)	50 µL x 100 µg/L	100 µL
0.10 µg/L (10 ppb)	100 µL x 100 µg/L	100 µL
0.20 µg/L (20 ppb)	200 µL x 100 µg/L	100 µL
0.50 µg/L (50 ppb)	500 µL x 100 µg/L	100 µL
1.00 µg/L (100 ppb)	100 µL x 1000 µg/L	100 µL
2.00 µg/L (200 ppb)	200 µL x 1000 µg/L	100 µL
5.00 µg/L (500 ppb)	500 µL x 1000 µg/L	100 µL
10.00 µg/L (1000 ppb)	1000 µL x 1000 µg/L	100 µL

* Equivalent Analyte concentration in a sample in ppb, assuming a sample concentration of 0.01 g/mL (0.5 g/50 mL) in final extract.

c. Quality Control Standards

Prepare Quality Control standards from commercially available standard solution(s), e.g. 10,000 µg/L, obtained from a *different source* than that used to prepare Calibration Standards. Two types of quality control standard must be prepared:

i. QC Standard:

Prepare a combined metals standard having concentrations near the midpoint of the calibration curve, but different from those used in any calibration standard. Prepare in same manner calibration standards are prepared, diluting with 2% HNO₃ and adding sufficient 5000 µg/L ISTD to result in a final ISTD concentration of 5 µg/L.

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ii. Intermediate Fortification Standard:

Use 10,000 µg/L metals standard solutions (D.2.b.i) to prepare an intermediate multimetal fortification standard with concentrations about 500 times the MLA, diluting to volume with 2% HNO₃.

Table 7 - Intermediate Fortification Standard Solution (IFSS) in 2% HNO₃

Analyte	Standard Concentration (µg/mL)	Volume (µL/100mL)	IFSS Concentration (ng/mL)
Pb	10000	125	12500
Cd	10000	50	5000
Co	10000	125	12500
Mo	10000	250	25000
Tl	10000	250	25000
Mn	10000	1000	100000
Se	10000	2500	250000

iii. Fortification Standard:

Place 500 µL of the intermediate fortification standard in a 50 mL polymeric volumetric flask and dilute to volume with 2% HNO₃. 100 µL of the Fortification Standard added to control tissue is equivalent to the MLA of the metals (100 µL = 25 ppb Pb, 10 ppb Cd, 25 ppb Co, 50 ppb Mo, 50 ppb Tl, 200 ppb Mn, and 500 ppb Se) based on a 0.5 g sample weight. Note: Standards containing other concentration ratios of metals may be used if desired.

d. Intermediate Stock/Fortification Standard - ICP-OES

- i. Prepare intermediate standard by making dilutions of commercially available standard solutions into a concentration of HNO₃ similar to the commercial solution.

Table 8 - Intermediate Standard/Fortification Concentrations - ICP-OES

Analyte	Standard Concentration (µg/mL)	Volume (µL/50mL)	Intermediate Std Concentration (µg/mL)
Fe	10,000	250	50.0
Zn	10,000	250	50.0
Cu	10,000	25	5.00

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Ni	10,000	50	10.0
Al	10,000	200	40.0
B	10,000	40	8.00
Ba	10,000	30	6.00
Cr	10,000	30	6.00
V	10,000	30	6.00
Sr	10,000	25	5.00

e. Calibration Standards - ICP-OES

Calibration standards are required for constructing a multipoint standard curve covering the range of analyte concentrations anticipated in samples.

Prepare calibration standards by making appropriate dilutions of intermediate standard in 2% HNO₃, adding 50 µL of 5000 µg/L ISTD, and dilute to 50 mL. Final ISTD concentration is 5 µg/L. Prepare these standards using polymeric volumetric flasks.

The Table below lists some suggested concentrations for calibration standards and recommended volumes of solutions required for preparation of 50mL volumes of each.

Table 9 - Calibration Standards - ICP-OES

Calibration STD Conc.	Amount used of Intermediate Standard in Table 8	Amount ISTD
Calibration Blank	2% HNO ₃ Only	50 µL
0.40 µg/mL	400 µL	50 µL
0.60 µg/mL	600 µL	50 µL
0.80 µg/mL	800 µL	50µL
1.0 µg/mL	1000 µL	50 µL

f. Quality Control Standards - ICP-OES

Prepare Quality Control standard(s) from commercially available standard solution(s), from a *different source* than that used to prepare Calibration Standards (Inorganic Ventures IV-19 & IV-7, 100 µg/mL, e.g.)

QC Standard - Calibration Curve Verification (CCV):

Prepare a combined metals standard having concentrations near the midpoint of the calibration curve, but different from those used in any calibration standard. Prepare in same manner calibration standards are prepared, diluting with 2% HNO₃ and adding sufficient 5000 µg/L ISTD to result in a final ISTD concentration of 5 µg/L.

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E. SAMPLE PREPARATION

Note: Since trace amounts of analyte metals are ubiquitous in the environment and may be present in dust particles, efforts should be made to avoid external contamination. All areas/materials involved in sample preparation and analysis should be kept as dust-free as possible to minimize the chance of contamination.

Samples must be thoroughly blended to assure uniformity prior to removal of a test portion.

F. ANALYTICAL PROCEDURE

1. Preparation of Controls and Samples

- a. Weigh homogenized samples, and blank tissue sufficient to prepare the negative control(s), positive control(s), and check samples (approximately 0.5 g for muscle tissues, 0.5 - 1 g for liver and kidney¹) for both ICP-MS and ICP-OES to the nearest 0.01 g into a clean² microwave vessel liner, using separate vessels for ICP-MS and ICP-OES controls. Teflon or polypropylene stirring rods may be used to manipulate samples.

Note: Truly blank tissues may not be available. Use previously analyzed tissues having low analyte levels for this purpose.

¹Caution! Mixing sample types or sample weights may produce unacceptably large variations in pressures developed during digestion, possibly resulting in damage to vessels if unvented caps are used. In order to maintain relatively constant digestion conditions in all unvented vessels, analyst should digest like quantities of similar sample matrices in each batch.

²Vessel liners, caps, stoppers, and tubing must be cleaned after each use to reduce the possibility of cross-contamination. Refer to Section J.2 for recommended cleaning procedure.

- b. Prepare positive control(s) by adding Fortification Standard to the tissue blank within the quantitation range. e. g.
 - i. ICP-MS: fortify with 100 µL of D.2.c.iii.
 - ii. ICP-OES: fortify with 300 µL of D.2.d.i.

2. Extraction Procedure

- a. Microwave Digestion
 - i. Add 5 mL of concentrated HNO₃, Ultra-pure grade, to each vessel.
 - ii. Assemble the vessel according to the manufacturer's instructions.
 - iii. Place assembled vessels into the microwave according to the manufacturer's instructions.

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- iv. Program oven with parameters demonstrated to safely and effectively digest samples (producing a clear digest when diluted). Recommended parameters are listed in the table below. It may be necessary to adjust these parameters to accommodate variations between individual instruments.

Power:	1200 Watts*
Ramp time:	10 minutes
Final temperature:	180 °C
Temperature hold time:	10 minutes
Cool down time:	10 minutes

*If there are less than eight vessels in the microwave the wattage can be lowered.

- v. Initiate oven program and digest samples.
- vi. Allow vessels to cool, then transfer to a fume hood and allow vessels to equilibrate to room temperature.
- vii. Slowly open the vent fittings and vent to atmospheric pressure, then disassemble vessels.
- b. Microwave Evaporation
- i. Place vessel liners into the evaporation carousel and assemble according to the manufacturer's instructions. Note: Manufacturer recommends use of PTFE syringe filters with the evaporation manifold for trace metal analysis (optional).
- ii. Place the evaporation assembly into the microwave.
- iii. Program oven to reduce solution volumes to approximately 1 mL. Typical program parameters are listed below.

Power:	600 Watts
Ramp time:	5 minutes
Final temperature:	120 °C
Temperature hold time:	3.5 minutes*
Cool down time:	10 minutes

*Typical value required when 8 vessels are used. Hold times required to achieve a final volume of 1 mL for any given number of vessels must be determined experimentally.

Note: If the microwave is capable of determining an evaporation plateau temperature, a temperature drop of $\Delta T \approx 7 \text{ }^\circ\text{C}$ may be used to control the

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final volume. This approach is more variable, but does not require adjustment for tissue type, tissue weight, or number of samples.

- iv. Initiate oven program and evaporate samples.
 - v. Once vessels have cooled to room temperature, remove the evaporation assembly from the oven and dismantle. Flush the evaporation manifold with DI water.
- c. Extraction Preparation
- i. If the solution volume remaining in the vessel liner is <1 mL, add concentrated HNO₃ to bring volume to approximately 1 mL. Note: residual acid volumes of up to 2.5 mL are acceptable, but should be avoided if possible. Pour extract solution into a 50 mL plastic tube containing approximately 10 mL millipure water.
 - ii. Quantitatively transfer residual digest by rinsing the liner 3 - 4 times with millipure water, adding each rinse to the extract in the tube. Keep total rinse volume < 35 mL.

Add 50 µL of 5000 µg/L ISTD solution to the extract.

Note: ISTD may be added online by the instrument.
 - iii. Bring extract volume to 50 mL with millipure water.
 - iv. Cap tube and invert several times to mix.

Note: The percentage of dissolved solids in the 50 mL extract, which is higher than that recommended by instrument manufacturer, can be reduced by increasing the dilution volume. Analyst must balance detrimental effects of high dissolved solids content (matrix effects, instrument contamination) against detrimental effects resulting from environmental contamination and lower analyte concentrations when considering this. If additional dilutions are made, care must be taken to maintain acid strength at ~2% and ISTD concentration at 5 µg/L. Adjust standard curve concentrations accordingly, if necessary.
 - v. Place sample and QC solutions on autosampler for analysis.

3. Instrumental Settings - ICP-MS Analysis

Note: The instrument parameters may be optimized to ensure system suitability.

a. Tuning

- i. Prior to sample analysis check the instrument's tuning parameters by analyzing the Mass Spectrometer Tuning Solution as specified by the manufacturer. Check the sensitivity, % RSD, % oxide, % doubly charged, peak shape, and resolution.

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b. ICP-MS Parameters

Set up instrument to monitor isotopes of appropriate metals, internal standards and modes.

Table 10. ICP- MS Metal Isotopes

Metal	Isotope <i>*Isotope used for quantitation</i>	Gas Mode
Manganese	55*	He
Cobalt	59*	He
Selenium	78*, 82	H ₂
Molybdenum	95, 98*	He
Cadmium	111*, 112,114	He
Thallium	203, 205*	He
Lead	206*, 207*, 208*	He

Note: These metals can be analyzed in no gas mode (no reaction cell), or another gas mode, but may require the use of interference correction equation(s).

Table 11. ICP-MS Internal Standard Isotopes

Internal Standard	Isotope
Scandium	45
Gallium	69, 71
Germanium	72
Yttrium	89
Rhodium	103
Indium	115
Terbium	159
Bismuth	209

Note: These metals can be analyzed using alternate internal standards/isotopes as necessary.

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c. Instrument calibration

- i. Analyze a calibration blank followed by at least 3 calibration standards (D.2.b) covering the range of interest. Using linear regression analysis, plot relative response (response relative to ISTD response) vs. concentration in $\mu\text{g/L}$ and determine slope (m), intercept (b), and correlation coefficient (r) of the calibration curve. This can be automatically performed by the ICP-MS software. Correlation coefficient r must be ≥ 0.995 , or calibration must be repeated.
- ii. Analyze the calibration blank and a QC standard (D.2.c.i.) immediately after the calibration curve. The response of the blank should be similar to that observed when initially analyzed. The calculated metals concentrations in the QC standard must be within $\pm 12\%$ of their accepted value. If these conditions are not met, the calibration sequence must be repeated until results are acceptable.

4. Instrumental Settings - ICP-OES Analysis

Note: The instrument parameters may be optimized to ensure system suitability.

a. Mn check

- i. Prior to sample analysis check the instrument's stability by analyzing a 1 ppm Mn solution and collecting 10 replicates. Check the sensitivity, and % RSD should be $\leq 2\%$.
- ii. If these parameters are outside the manufacturer's specifications or QC requirements, examine pump tubing, drain tubing, torch condition, etc., and then reanalyze check solution.

b. ICP-OES Parameters

Set up instrument to monitor responses of appropriate metals, and selected internal standards.

Table 12. ICP-OES Wavelengths

Metal	Analytical Wavelength	Internal Std
Iron	239.562	Scandium
Zinc	213.857	Scandium
Copper	324.752	Yttrium
Nickel	231.604	Yttrium
Aluminum	396.153	Yttrium
Boron	249.772	Yttrium
Barium	233.527	Yttrium

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Metal	Analytical Wavelength	Internal Std
Chromium	267.716	Yttrium
Vanadium	292.402	Yttrium
Strontium	460.733	Yttrium

Note: These metals can be analyzed using alternate wavelengths and/or internal standards as necessary.

Table 13. ICP-OES Internal Standard Wavelengths

Internal Standard	Wavelength
Scandium	361.383
Gallium	294.364
Germanium	209.426
Yttrium	371.029
Rhodium	233.477
Indium	230.606
Terbium	350.917
Bismuth	223.061

- c. Instrument calibration
- i. Analyze a calibration blank followed by at least 3 calibration standards (D.2.e) covering the range of interest. Using linear regression analysis, plot relative response (response relative to ISTD response) vs. concentration in $\mu\text{g/mL}$ and determine slope (m), intercept (b), and correlation coefficient (r) of the calibration curve. This can be automatically performed by the ICP-OES software. Correlation coefficient r must be ≥ 0.995 , or calibration must be repeated.
 - ii. Analyze the calibration blank (optional) and a QC standard (D.2.f.) immediately after the calibration curve. The relative response of the blank should be similar to that observed when initially analyzed. The calculated metals concentrations in the QC standard must be within $\pm 12\%$ of their accepted value. If these conditions are not met, the calibration sequence must be repeated until results are acceptable.

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5. Injection sequence / Sample Set (ICP-MS and ICP-OES)
 - a. Calibration blanks (optional for ICP- OES) and QC standards must be included in the sample analysis sequence to verify instrument performance over the course of the run.
 - b. If response of any sample exceeds highest standard in the calibration curve, make an appropriate dilution in 2% HNO₃ and add ISTD to maintain a 5 µg/L concentration, then re-analyze.
 - c. A sample set for both ICP-MS and ICP-OES consists of the following:

Note: Each sample set must contain one QA sample/20 samples.

 - i. Reagent Blank (optional)
 - ii. Negative control.
 - iii. Positive control(s).
 - iv. Samples.

G. CALCULATIONS / IDENTIFICATION (ICP-MS and ICP-OES)

Note: Instrument software can be programmed to perform all necessary calculations.

1. Using values for m, b determined for the calibration curve (F.3.c), determine selected analyte concentration (C_E, in µg/L) in any extract having a relative response R using:

$$C_E (\mu\text{g/L}) = C_E, \mu\text{g/L} = (R-b)/m$$

Note: If sample is found to contain molybdenum, instrument software must be set to compensate for contribution of molybdenum oxide to the 111 isotope used for quantitation of cadmium in the sample.

2. Calculate selected analyte concentrations in digested controls and samples (C_S) using:

$$C_S (\text{ppb}) = \frac{C_E \times V_E \times D}{W}$$

Where

C_E = Analyte concentration in final extract, in µg/L

V_E = Final sample extract volume in milliliters

D = Dilution factor (Diluted volume/aliquot volume), if secondary dilution was made.

W = Sample Weight in grams.

3. Calculate Relative % Difference (RPD) for duplicate results using:

$$\text{RPD} = \frac{|C_1 - C_2| \times 200}{(C_1 + C_2)}$$

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Where:

C1 = first duplicate's concentration.

C2 = second duplicate's concentration.

4. Calculate recoveries of fortified controls and check samples using

$$\%Rec = \frac{(C_F - C_B) \times W \times 100}{V_{FS} \times C_{FS}}$$

Where

C_F, C_B = Analyte concentrations determined for the fortified sample and the blank tissue from which it was prepared, in ppb (ng/g).

W = Weight of fortified control, in grams.

V_{FS} = Volume of fortification standard added, in mL.

C_{FS} = Concentration of fortification standard, in µg/L.

H. SAFETY INFORMATION AND PRECAUTIONS

1. Required Protective Equipment - Safety glasses, lab coat, protective gloves.
2. Hazards

<i>Procedure Step</i>	<i>Hazard</i>	<i>Recommended Safe Procedures</i>
Nitric Acid	Strong oxidizer. May be fatal if swallowed or inhaled. Extremely corrosive. Contact with skin or eyes may cause severe burns and permanent damage.	Perform operations using concentrated acid in fume hood. Use protective eyewear, gloves and clothing. Store in approved acid safety cabinet away from basic or other reactive materials.
Microwave Digester	Possible explosion hazard	Follow manufacturer recommendations
Metals Standards	Poisonous if ingested.	Do not pipet by mouth

3. Disposal Procedures
Follow local, state and federal guidelines for disposal.

I. QUALITY ASSURANCE PLAN

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1. Performance Standard

Table 14 - Performance Standard ICP-MS

<i>Analyte</i>	<i>Analytical Range (ppb)</i>	<i>Acceptable Recovery (%)</i>
Cadmium	≥ 10	71-132
Lead	≥ 25	81-127
Selenium	≥ 500	72-146
Manganese	≥ 200	81-110
Molybdenum	≥ 50	90-115
Thallium	≥ 50	95-118
Cobalt	≥ 25	86-107

Table 15 - Performance Standard ICP-OES

<i>Analyte</i>	<i>Analytical Range (ppm)</i>	<i>Acceptable Recovery (%)</i>
Iron	≥ 30	85 - 115
Zinc	≥ 30	83 - 121
Copper	≥ 3.0	80 - 118
Nickel	≥ 6.0	88 - 110
Aluminum	≥ 24	86 - 120
Boron	≥ 4.8	80 - 109
Barium	≥ 3.6	87 - 116
Chromium	≥ 3.6	89 - 109
Vanadium	≥ 3.6	90 - 111
Strontium	≥ 3.0	94 - 153

For each sample set:

- a. The instrument calibration meets specifications in section F.3.c. and F.4.c.
- b. For set acceptance, 90% of the monitored analytes in the positive control must meet specifications. To report a positive sample, the analytes of interest in the positive control must meet specifications.
- c. If a positive control duplicate is run, the calculated RPD is ≤ 20%.
- d. All calibration blanks injected show consistent responses, and the first QC standard is within ± 12% of the accepted value.

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- e. For ICP-MS perform the following: for each sample within the set, the internal standard response is within $\pm 50\%$ of the average instrument calibration internal standard response.

2. Critical Control Points and Specifications

None known

3. Intralaboratory Check Samples

- a. System, minimum contents.

- i. Frequency: One per week per analyst when samples analyzed.
- ii. Records are to be maintained.

- b. Acceptability criteria.

Refer to I. 1.

If unacceptable values are obtained, then:

- i. Investigate following established procedures.
- ii. Take corrective action as warranted.

4. Sample Condition upon Receipt: Cold

J. APPENDIX

1. References

- a. Agilent 7500 ICP-MS Hardware Manual, G1833-90004, January 2001.
- b. CEM XP-1500 Plus Vessel Accessory Sets and Autovent Option Instruction for Use, 600493, Rev. 5, 8/01.
- c. CEM Vacuum Concentration/Drying Accessory Set Instructions for Assembly and Use, 600484, Rev. 1, 6/99.
- d. CEM Mars Operation Manual, 600122, Rev 2, February, 2006.
- e. EPA Method 6020, Inductively Coupled Plasma-Mass Spectrometry, Revision 0, September 1994.
- f. Perkin Elmer 5000 Series Hardware Guide, 0993-6762, D, February 2007.

2. Cleaning Vessel Liners

The following procedures are suitable for removal of residual adsorbed residues from Teflon liners used in this method. Other procedures are available and may be used if demonstrated to be effective.

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Option 1

- a. Add approximately 10 mL of conc. HNO₃ (metal analysis grade) to each digestion vessel liner.
- b. Assemble vessels as specified by manufacturer.
- c. Place in microwave.
- d. Digest at 600W, ramp to 150 °C over 10 minutes, then hold at temperature for 10 minutes.
- e. Cool vessels to room temperature, then disassemble.
- f. Rinse vessel liners and caps with millipure water several times to remove all traces of acid.
- g. Place in a clean environment to dry.
- h. Reference: CEM VesselCleaning09.doc, 04/09

Option 2

Using a Milestone Trace Clean apparatus and conc. metal analysis grade HNO₃:

- a. Place the microwave vessel liners and caps into the apparatus.
- b. Start the method program as per the manufacturer suggestion.
- c. After the apparatus has cooled, remove the liners and caps.
- d. Rinse several times with Ultrapure water to remove all traces of acid.
- e. Place in a clean environment to dry.

K. APPROVALS AND AUTHORITIES

1. Approvals on file.
2. Issuing Authority: Director, Laboratory Quality Assurance Staff.